

MEMBRANE STRUCTURE AND FUNCTION

Membrane transport

“Got to get it there”

Chapter 7 Continued

Objectives

- Understand what factors influence membrane permeability
- Understand the difference between passive and active transport
- Be able to discuss the processes of: Diffusion, Osmosis, Facilitated diffusion Pumps, Cotransport, Exocytosis and Endocytosis

Why do materials move?

- Molecules are in motion
- The motion of molecules increases the entropy of the system (less order)
- Molecular collisions result in molecules being moved along a gradient (concentration gradient or free energy gradient)
- Through random molecular collisions directional motion can be accomplished

Membrane Permeability

- Hydrophobic middle of the bilayer inhibits the passage of ions and polar molecules that are hydrophilic
- Transport proteins may assist ions and polar molecules across the membrane
- **Can cross** lipid bilayer = nonpolar, small uncharged polar molecules
- **Cannot** cross lipid bilayer = large polar and charged molecules

Transport Mechanisms

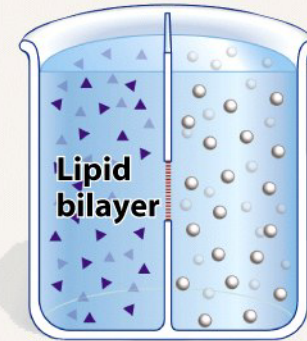
The movement of materials across membranes takes many routes but the mechanisms are categorized as either passive or active transport mechanisms

- **Passive Transport**
 - moves molecules along a concentration gradient
 - no cellular energy required
- **Active transport**
 - moves molecules against a concentration gradient
 - requires cellular energy

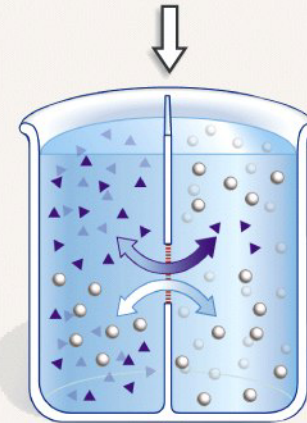
Passive Transport Mechanism

- **Simple Diffusion:** the movement of a substance from higher concentration to lesser concentration
- Simple diffusion occurs across the lipid bilayer
- The bilayer is selectively permeable as not everything can get across

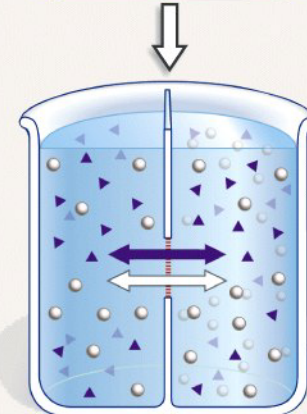
DIFFUSION ACROSS A LIPID BILAYER



1. Start with different solutes on opposite sides of a lipid bilayer. Both molecules diffuse freely across bilayer.



2. Solutes diffuse across the membrane — each along its own concentration gradient.



3. Equilibrium is established. Solutes continue to move back and forth across the membrane but at equal rates.

Specific examples of diffusion:

Osmosis

- **Osmosis:** the diffusion of water (solvent) across a membrane
 - influenced by total solute concentration
- **Osmotic pressure:** pressure exerted on a membrane due to an imbalance of solute between the inside and outside of the membrane
- ✓ **Water always moves toward the side with a greater concentration of solute**

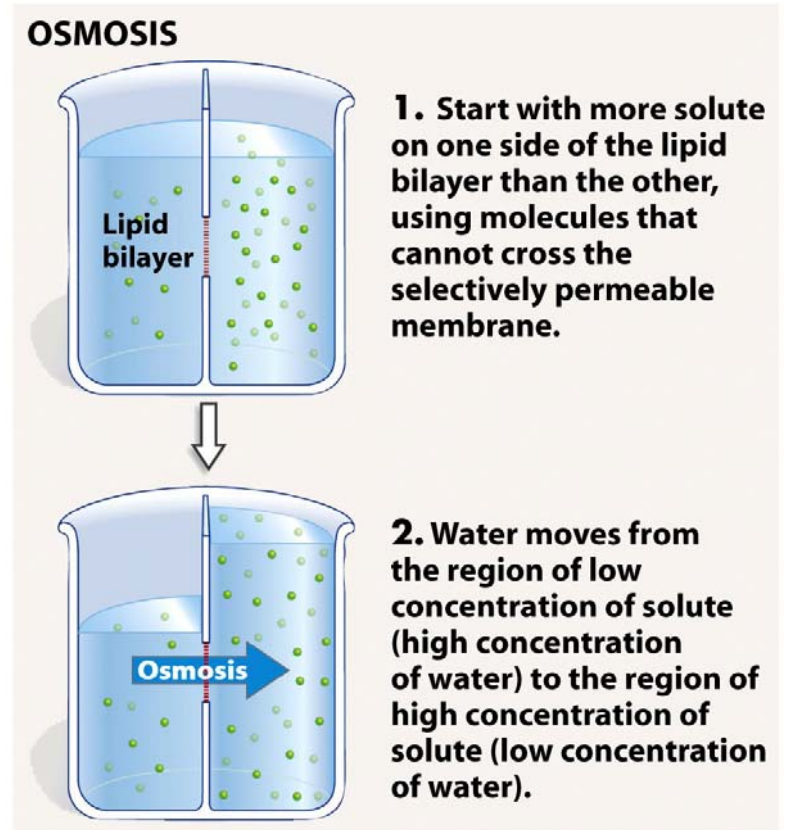
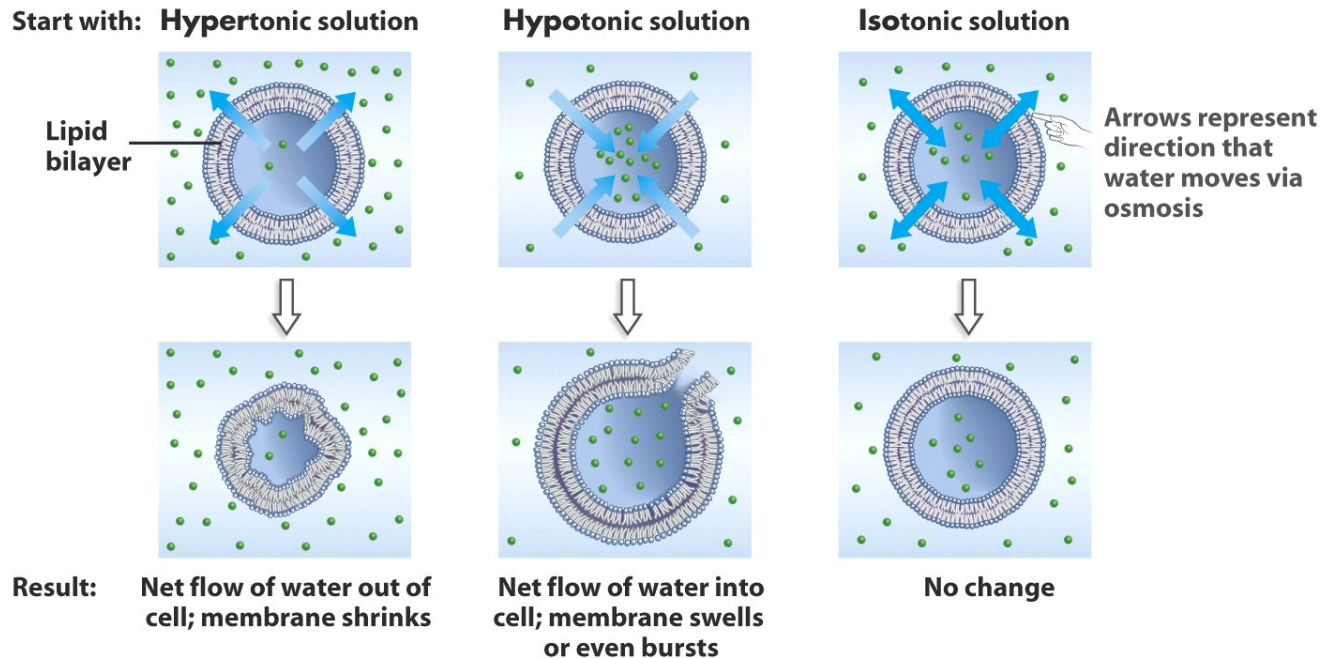


Figure 6-16 Biological Science, 2/e

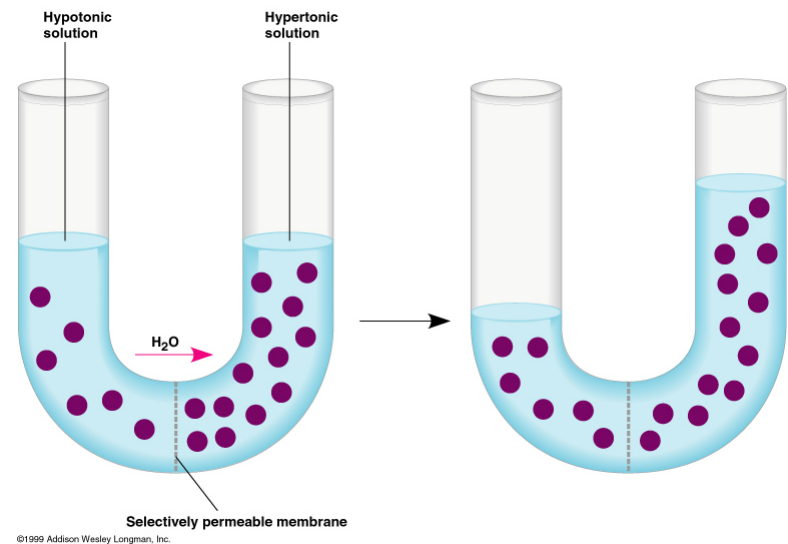
Cells respond to osmosis, so what?



- **Tonicity:** the ability of a solution to move water
 - Hypertonic: Greater ability to move H_2O ; gains water
 - Hypotonic: Lesser ability to move H_2O ; loses water
 - Isotonic: equal ability to move H_2O ; no net water movement

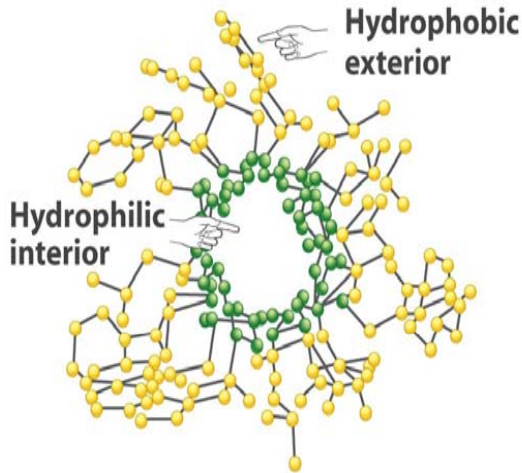
The Importance of Osmoregulation

- Living things must balance water uptake and loss
- If cells lose water they crenate
- If cells gain water they lyse
- Expulsion vacuole
- Turgor pressure

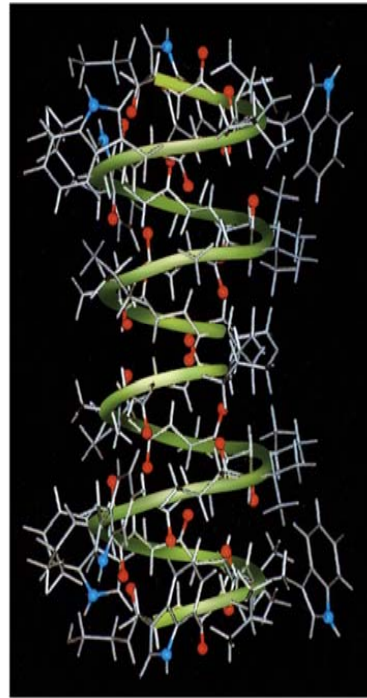


Transport Proteins: Facilitated Diffusion Via Channel Proteins

(a) Top view of gramicidin

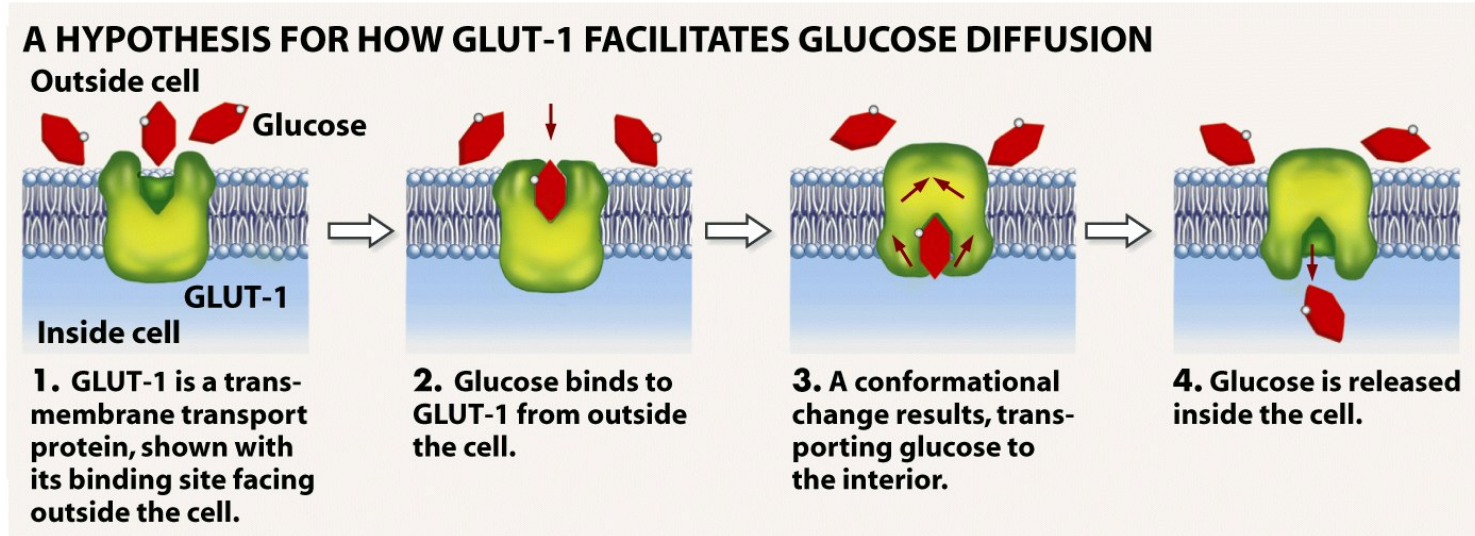


(b) Side view of gramicidin



- Involves transport proteins moving a solute along a concentration gradient
- May be specific
- May be saturated (can only work so fast) or inhibited
- Passive mechanism

Transport Proteins: Facilitated Diffusion Via Carrier Proteins



- Molecule causes a controlled denaturation resulting in a molecule being transported
- May be specific
- May be saturated or inhibited
- Protein assists the process of diffusion; passive mechanism

Regulation of Facilitated Diffusion

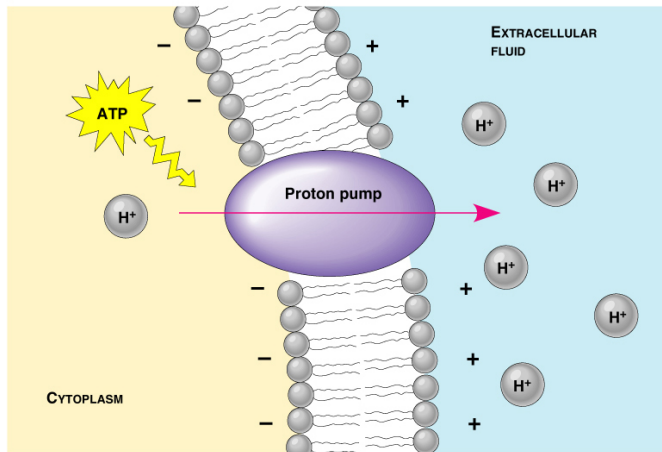
- Some transport proteins are regulated by chemical or electrical stimuli
- Usually these proteins are permanent channels that are opened or closed via other proteins (gated channels)

Active Transport: Pumps

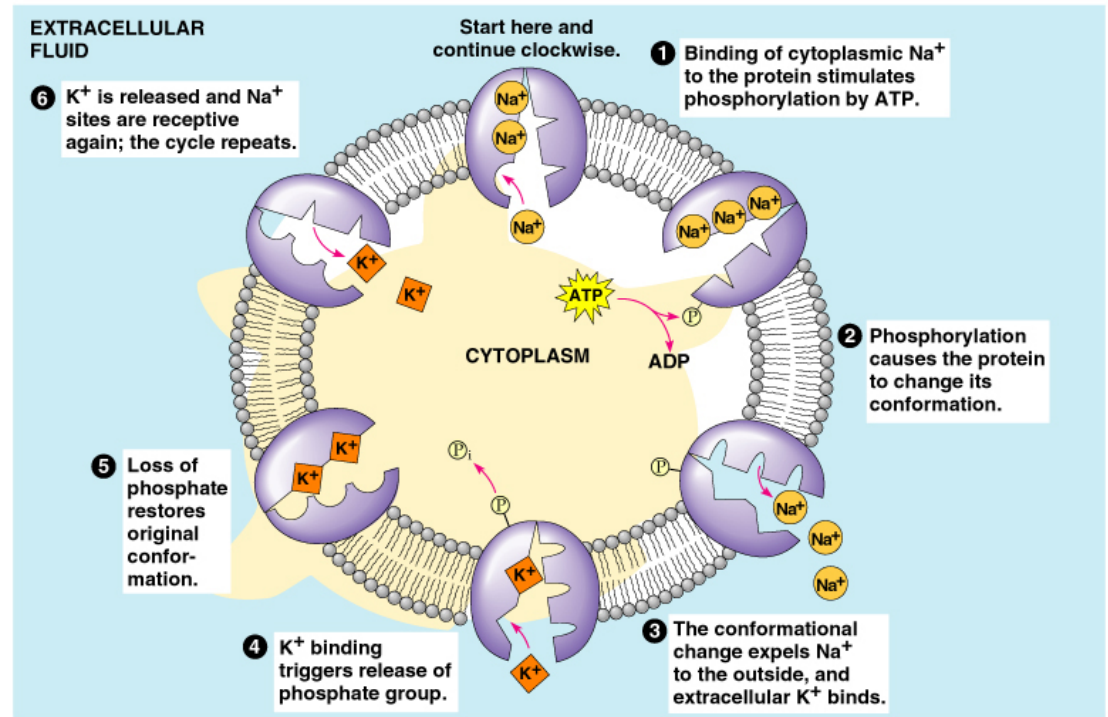
- Moves solute uphill and requires energy
- Always requires carrier proteins
- Major factor that allows the cell to regulate the concentration of solute within the cell
- May result in an **imbalance of solute** across a membrane that the cell can utilize

Kinds of Pumps

- Na^+/K^+
- H^+
- Ca^{2+}



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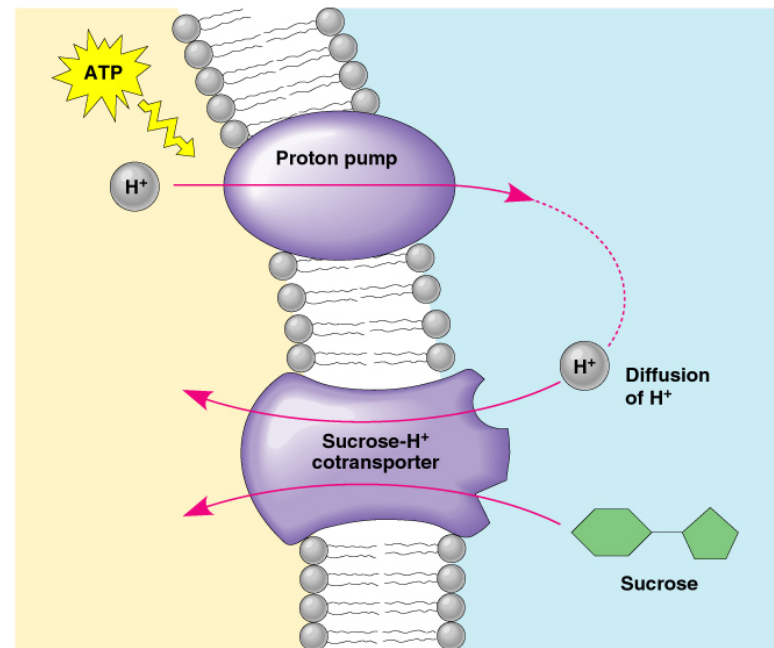
Some pumps create electrical differences across a membrane (electrogenic pumps)

Electrochemical Gradient

- The difference in voltage across a membrane resulting from electrogenic pumps is called **membrane potential**
- This electrical force affects the transport of charged solutes
- Cations are favored because interior of cell is usually negative compared to the outside
- Resulting **Electrochemical Gradient** affects ion transport:
 - electrical = membrane potential
 - chemical = concentration gradient

Cotransport (secondary active transport)

- Two solutes transported at one time via a transport protein
 - one with a gradient
 - one against a gradient
- Solute moving with the gradient does so because of an earlier active transport event (pump)



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Exocytosis and Endocytosis

- Exocytosis involve the movement of macromolecules out of the cell by the fusion of membrane bound vesicles to the plasma membrane
- Endocytosis involves the movement of macromolecules into the cell by the pinching of the plasma membrane into membrane bound vesicles
 - Phagocytosis
 - Pinocytosis
 - Receptor-mediated pinocytosis

Endocytosis

- Phagocytosis: ingestion of large particle
- Pinocytosis: ingestion of small mixed solutes
- Receptor-mediated pinocytosis: ingestion of specific solutes (ligands) with the aid of binding proteins; areas called coated pits

